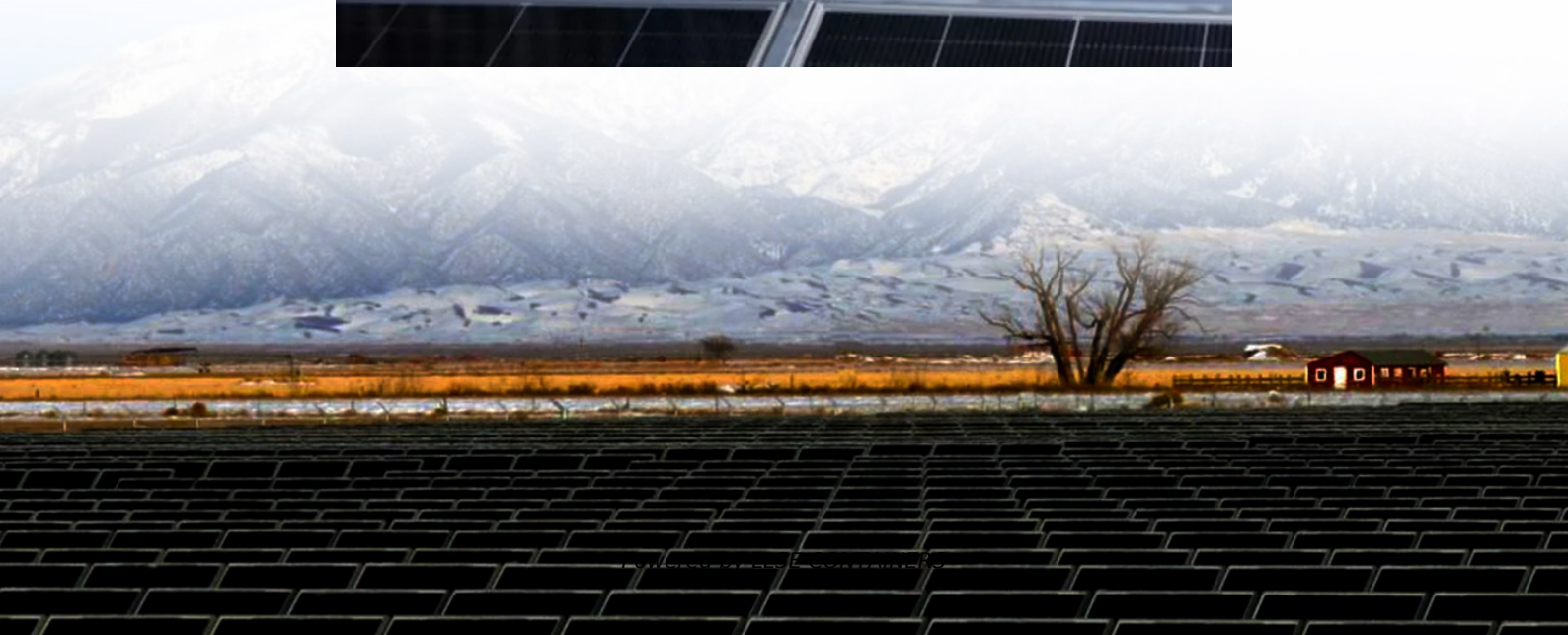


# **Fully Antioxidant Redox Flow Battery**





## Overview

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Flow batteries are one option for future, low-cost stationary energy storage. We present a perspective overview of the potential cost of organic active materials for aqueous flow batteries based on a comprehensive mathematical model. Flow batteries are one option for future, low-cost stationary energy storage. We present a perspective overview of the potential cost of organic active materials for aqueous flow batteries based on a comprehensive mathematical model. The battery capital costs for 38 different organic active materials, as well as the state-of-the-art vanadium system.

With an increasing focus on renewable energy resources, the search for economic stationary energy storage systems is more important than ever<sup>1,2</sup>. One promising electrochemical storage technology is the redox flow battery (RFB) in which the charge carriers are stored in liquid electrolytes and pumped through an electrochemical cell referred to as flow cell. This open cell architecture allows to decouple the place of the electrochemical reaction from the place where the energy is stored; peak power and capacity can be scaled independently from each other<sup>3</sup>. Thus, RFBs are very versatile and can be applied in different ranges of applications<sup>4</sup>. Additionally, in the case where aqueous electrolytes are applied, safety concerns are low in comparison to other energy storage systems like the lithium-ion battery.

Based on the model described in the section Method we calculated the capital costs of 38 organic active materials. In the following, we present the obtained results and break down the individual factors that contribute to the overall costs of an RFB with the respective active material class. To adjust for possible changes in costs due to possible optimization states of the RFB system, we discuss the results for both the AqORFB and the VRFB by means of two self-defined scenarios (cf. Table 1), with: (a) “Present Case”, using state-of-the-art values as reported in literature or given by industry/companies. This choice implies that we apply an estimated material price at the present moment given by literature. The Nafion membrane is selected as separator material. Further semipermeable material.

The calculations for our study are based on a model proposed by Brushett et al.<sup>14</sup>, a further developed TE model by Dmello et al.<sup>35</sup>, which is an extension of the approach by Darling et al.<sup>36</sup>. Additionally, we combine the aforementioned studies with reasonable assumptions for the size and costs of redox flow stack components by Minke et al.<sup>5,37</sup>. Furthermore, we extend the TE model approach with a more detailed polarization model that incorporates



kinetic as well as transport phenomena to account for individual differences in the physico-chemical properties of the active material. This section provides an overview of the calculation model used in our study. For an in-depth description, we refer the reader to section 1 in Supplementary Information.We.

All data generated in this study are included in the manuscript and Supplementary Information file. Source data are provided in this paper.

What are aqueous organic redox flow batteries?

Recently, aqueous organic redox flow batteries (AORFBs), utilizing water-soluble organic molecules as redox-active species, have garnered widespread attention [8, 9]. The conversion between electrical and chemical energy in organic molecules often involves electron transfer at active centers such as oxygen, nitrogen, sulfur, or radicals, etc.

Can organic redox-active materials be used for Advanced Flow batteries?

Organic redox-active materials offer a new opportunity for the construction of advanced flow batteries due to their advantages of potentially low cost, extensive structural diversity, tunable electrochemical properties, and high natural abundance.

Can redox flow batteries be used for energy storage?

Ye, R. et al. Redox flow batteries for energy storage: a technology review. J. Electrochem. Energy Convers. Storage 15, 10801–10802 (2018). Gregory, T. D., Perry, M. L. & Albertus, P. Cost and price projections of synthetic active materials for redox flow batteries. J. Power Sources 499, 229965 (2021).

Can redox flow batteries be membrane-free?

Nonaqueous redox flow batteries face challenges like costly membranes and unstable electrolytes. Here, authors develop a membrane-free battery using a polypropylene carbonate gel polymer electrolyte with Li anode and Tri-TEMPO catholyte, achieving a high voltage of 3.45 V, capacity retention of 96.8%, and efficiency of 98.4%.



## Fully Antioxidant Redox Flow Battery

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### Redox Flow Battery

The main difference with conventional batteries is that the energy and power of the redox-flow battery are fully independent: the energy is related to the electrolyte volume (tank size) and the ...



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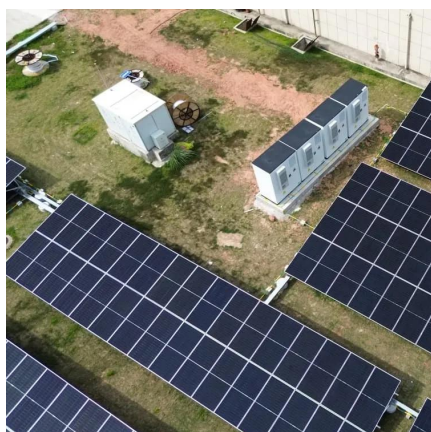
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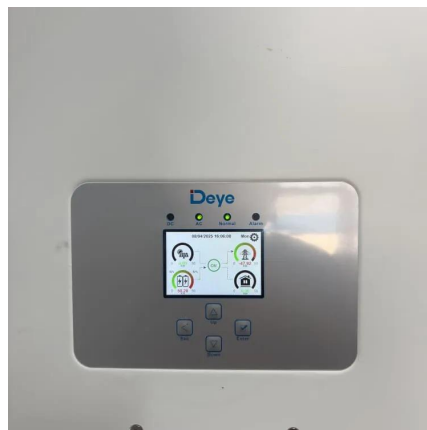
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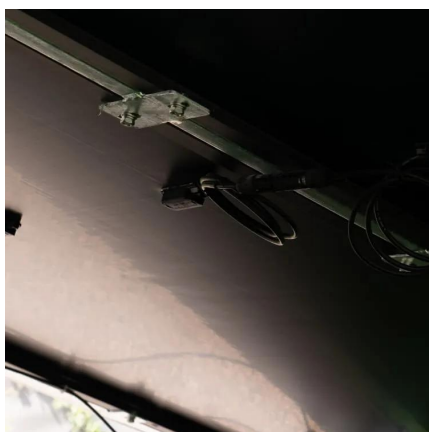
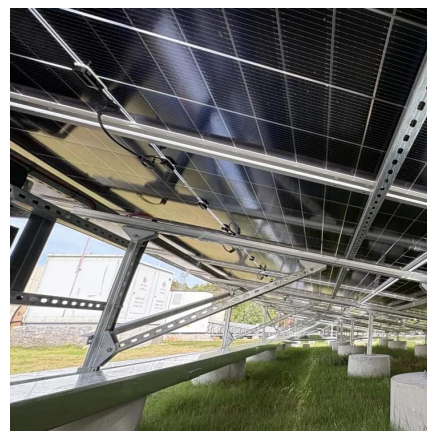


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### **Redox Flow Battery**

10.17.3 Redox flow batteries The redox flow batteries are flow batteries that employ two fully soluble redox couple solutions in each half-cell. Unlike the Zn/Br flow battery, the redox flow ...





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